

Cambridge International Examinations Cambridge International Advanced Subsidiary and Advanced Level

CHEMISTRY

9701/35 May/June 2017

Paper 3 Advanced Practical Skills 1 CONFIDENTIAL INSTRUCTIONS

Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly.



The Supervisor's attention is drawn to the Supervisor's Report on page 7 which must be completed and returned with the scripts.

If you have any queries regarding these Confidential Instructions, please contact Cambridge stating the Centre number, the nature of the query and the syllabus number quoted above.

email info@cie.org.uk phone +44 1223 553554 fax +44 1223 553558

This document consists of 8 printed pages.



Safety

2

Supervisors are advised to remind candidates that **all** substances in the examination should be treated with caution.

Only those tests described in the Question Paper should be attempted.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out. The following codes are used where relevant.

- C corrosive MH mod
- **HH** health hazard**F** flammable

- MH moderate hazard
- T acutely toxicO oxidising
- flammable O
- **N** hazardous to the aquatic environment

The attention of Supervisors is drawn to any local regulations relating to safety and first aid.

Hazard Data Sheets should be available from your chemical suppliers.

Before the examination

1 Access to the Question Paper is NOT permitted in advance of the examination.

2 Preparation of materials

Where quantities are specified for each candidate, they are sufficient for the experiments described in the Question Paper to be completed.

In preparing materials, **the bulk quantity for each substance should be increased by 25%** as spare material should be available to cover accidental loss. More material may be supplied if requested by candidates, without penalty.

All solutions should be bulked and mixed thoroughly before use to ensure uniformity.

Every effort should be made to keep the concentrations accurate.

If the concentrations differ slightly from those specified, the Examiners will make the necessary allowance. They should be informed of the exact concentrations.

3 Labelling of materials

Materials must be labelled as specified in these Confidential Instructions. Materials with an **FA** code number should be so labelled **without** the identities being included on the label. Where appropriate the identity of an **FA** coded chemical is given in the Question Paper itself.

4 Identity of materials

It should be noted that descriptions of materials given in the Question Paper may not correspond with the specifications in these Confidential Instructions. **The candidates must assume the descriptions given in the Question Paper.**

5 Size of group

In view of the difficulty of the preparation of large quantities of solution of uniform concentration, it is recommended that the maximum number of candidates per group be 30 and that separate supplies of solutions be prepared for each group.

Apparatus

- 1 In addition to the fittings ordinarily contained in a chemical laboratory, the apparatus and materials specified below will be necessary.
- 2 Pipette fillers (or equivalent safety devices), suitable eye protection and disposable gloves should be used where necessary.
- **3** For each candidate
 - $2 \times 50 \text{ cm}^3$ burette (one of these should be labelled **FA 1**)
 - $2 \times$ burette stand and clamp
 - 2 × funnel (for filling burette)
 - $1 \times 25 \text{ cm}^3$ measuring cylinder
 - $2 \times 100 \, \text{cm}^3$ beaker
 - 1 × glass rod
 - $1 \times$ white tile
 - $5 \times \text{test-tube}$
 - 2 × boiling tube*
 - 1 × test-tube rack
 - 1 × test-tube holder
 - 3 × teat/dropping pipette
 - $1 \times spatula$
 - $1 \times Bunsen burner$
 - $1 \times heatproof mat$
 - $1 \times$ wash bottle containing distilled water
 - 1 × pen (suitable for labelling glassware)
 - 1 × stop clock or sight of a clock with seconds display
 - paper towels

*Candidates are expected to rinse and reuse test-tubes and boiling tubes where possible. Additional tubes should be available. **Chemicals required**

- It is especially important that great care is taken that the confidential information given below does not reach the candidates either directly or indirectly.
- It should be noted that descriptions of substances given in the Question Paper may not correspond with the specifications in these Confidential Instructions. 2
- 3 Particular requirements

Image:	hazard	label	per candidate	identity	notes (hazards given in this column are for the raw materials)
FA2150 cm³1.0 mol dm⁻³ potassium iodideFA3100 cm³0.0050 mol dm⁻³ sodiumFA3100 cm³0.0050 mol dm⁻³ sodiumstarch indicator15 cm³starch indicatorstarch indicator15 cm³starch indicatorFA420 cm³0.2 mol dm⁻³ ammoniumFA520 cm³0.2 mol dm⁻³ potassiumFA60.5gsodium sulfateFA710 cm³1.0 mol dm⁻³ sulfuric acidMagnesium1 × 2 cm1.0 mol dm⁻³ sodium nitritedistilled water100 cm³distilled water		FA 1	120 cm ³	0.020 mol dm ⁻³ potassium peroxodisulfate	Dissolve 5.41 g of $K_2S_2O_8$ [HH] [MH] [O] in each dm ³ of solution.
FA 3100cm³0.0050 moldm⁻³ sodiumstarch indicator15 cm³starch indicatorstarch indicator15 cm³starch indicatorFA 420 cm³iron(II) sulfateFA 520 cm³0.2 moldm⁻³ ammoniumFA 60.5gsodium sulfateFA 710 cm³1.0 mol dm⁻³ sodiumFA 710 cm³1.0 mol dm⁻³ sodiumFA 710 cm³1.0 mol dm⁻³ sodiumFA 810 cm³1.0 mol dm⁻³ sodium nitriteFA 810 cm³1.0 mol dm⁻³ sodium nitritefatilled water100 cm³distilled waterdistilled water100 cm³distilled water		FA 2	150 cm ³		Dissolve 166.0g of KI in each dm^3 of solution.
starch indicator15 cm³starch indicatorstarch indicator15 cm³starch indicatorFA 420 cm³0.2 mol dm³ ammoniumFA 520 cm³0.2 mol dm³ potassiumFA 60.5 gsodium sulfateFA 710 cm³1.0 mol dm³ sulfuric acidFA 810 cm³1.0 mol dm³ sulfuric acidFA 810 cm³1.0 mol dm³ sulfuric acidfate10 cm³1.0 mol dm³ sodium nitritedistilled water100 cm³distilled water		FA 3	100 cm ³	0.0050 mol dm ⁻³ sodium thiosulfate	Dissolve 24.8g of $Na_2S_2O_3.5H_2O$ in each dm ³ of solution. Dilute this 20-fold.
FA420cm³0.2 moldm⁻³ ammonium iron(II) sulfateFA520cm³0.2 moldm⁻³ potassium aluminium sulfateFA60.5gsodium sulfateFA710 cm³1.0 moldm⁻³ sulfuric acidFA810 cm³1.0 moldm⁻³ sulfuric acidFA810 cm³1.0 moldm⁻³ sulfuric acidfa810 cm³1.0 moldm⁻³ sodium nitritemagnesium1 × 2 cm stripmagnesium ribbondistilled water100 cm³distilled water		starch indicator	15 cm ³	starch indicator	See preparation instructions in the current syllabus.
FA 5 20 cm^3 $0.2 \text{ moldm}^{-3} \text{ potassium}$ FA 6 0.5 g $0.2 \text{ moldm}^{-3} \text{ moldm}^{-3} \text{ moldm}^{-3}$ FA 7 0.5 g $0.5 \text{ moldm}^{-3} \text{ moldm}^{-3} \text{ moldm}^{-3}$ FA 7 10 cm^3 $1.0 \text{ moldm}^{-3} \text{ sodium nitrite}$ FA 8 10 cm^3 $1.0 \text{ moldm}^{-3} \text{ sodium nitrite}$ magnesium $1 \times 2 \text{ cm}$ magnesium ribbondistilled water 100 cm^3 distilled water		FA 4	20 cm ³		Dissolve 78.4 g of $(NH_4)_2$ Fe $(SO_4)_2$.6H $_2$ O [MH] in each dm ³ of solution.
FA 60.5 gsodium sulfiteFA 70.5 gsodium sulfiteFA 710 cm³1.0 mol dm⁻³ sulfuric acidFA 810 cm³1.0 mol dm⁻³ sodium nitritemagnesium1 × 2 cmmagnesium ribbondistilled water100 cm³distilled water		FA 5	20 cm ³	0.2 mol dm ⁻³ potassium aluminium sulfate	Dissolve 94.8g of $KAl(SO_4)_2$.12 H_2O in each dm ³ of solution.
FA 710 cm³1.0 mol dm⁻³ sulfuric acidFA 810 cm³1.0 mol dm⁻³ sodium nitritemagnesium1 × 2 cmmagnesium ribbondistilled water100 cm³distilled water	[нн]	FA 6	0.5g	sodium sulfite	Each candidate should be provided with approximately 0.5g of Na_2SO_3 [HH] in a stoppered bottle.
FA 810 cm³1.0 mol dm⁻³ sodium nitritemagnesium1 × 2 cm stripmagnesium ribbondistilled water100 cm³distilled water	[MH]	FA 7	10 cm ³		See preparation instructions in the current syllabus.
magnesium $1 \times 2 \text{ cm}$ stripmagnesium ribbondistilled water 100 cm^3 distilled water	[HW]	FA 8	10 cm ³		Dissolve 69.0g of NaNO $_2$ [O] [T] [N] in each dm 3 of solution.
$100\mathrm{cm}^3$	E	magnesium	1 × 2 cm strip		More pre-cut strips should be available for candidates who wish to repeat the test.
		distilled water	100 cm ³	distilled water	

NOTE: Small amounts of SO₂ [C] [T] and NO₂ [C] [O] [T], which can cause respiratory distress in some people, may be produced. The laboratory must be well ventilated.

hazard	label	notes
	dilute hydrochloric acid	
[c]	dilute nitric acid	
[MH]	dilute sulfuric acid	
[MH]	aqueous ammonia	
[C]	aqueous sodium hydroxide	See identity details and preparation instructions in the current syllabus.
	0.1 moldm ⁻³ barium chloride or 0.1 moldm ⁻³ barium nitrate	
[N]	0.05 mol dm ⁻³ silver nitrate	
[MH]	limewater	
[MH]	aqueous acidified potassium manganate(VII)	

5 The following materials and apparatus should be available.

red and blue litmus papers, aluminium foil for testing nitrate/nitrite, wooden splints and the apparatus normally used in the Centre for use with limewater in testing for carbon dioxide

Responsibilities of the Supervisor during the examination

1 The Supervisor, or other competent chemist, **must**, **out of sight of the candidates**, **carry out the experiments in Question 1** and complete tables of readings on a spare copy of the Question Paper. This should be labelled 'Supervisor's Results' and show the Centre number and appropriate session/laboratory number.

This should be done for **each session** held and **each laboratory** used in that session, and **each batch** of solutions supplied.

N.B. The Question Paper front cover requests the candidate to fill in details of the examination session and the laboratory used for the examination.

It is essential that each packet of scripts contains a copy of the applicable Supervisor's Results as the candidates' work cannot be assessed accurately without such information.

2 The Supervisor must complete the Supervisor's Report on page 7 to show which candidates attended each session. If all candidates took the examination in one session, please indicate this on the Supervisor's Report. A copy of the Supervisor's Report must accompany each copy of the Supervisor's Results in order for the candidates' work to be assessed accurately.

The Supervisor must give details on page 8 of any particular difficulties experienced by a candidate, especially if the Examiner would be unable to discover this from the written answers.

After the examination

Each envelope returned to Cambridge must contain the following items.

- 1 The scripts of those candidates specified on the barcode label provided.
- **2** A copy of the Supervisor's Results relevant to the candidates in **1**.
- **3** A copy of the Supervisor's Report, including details of any difficulties experienced by candidates (see pages 7 and 8).
- 4 The Attendance Register.
- 5 A Seating Plan for each session/laboratory.

Failure to provide appropriate documentation in each envelope may cause candidates to be penalised.

Colour blindness

With regard to colour blindness it is permissible to advise candidates who request assistance on colours of, for example, precipitates and solutions (especially titration end-points). Please include with the scripts a note of the candidate numbers of such candidates.

Experience suggests that candidates who are red/green colour-blind – the most common form – do not generally have significant difficulty. Reporting such cases with the scripts removes the need for a Special Consideration Form.

SUPERVISOR'S REPORT

This form must be completed and returned in the envelope with the scripts, the Supervisor's Results, the Attendance Register and the Seating Plan.

Centre number	Name of Centre

The candidate numbers of candidates attending each session were:

first session second session

The Supervisor is required to give details overleaf of any difficulties experienced by particular candidates, giving names and candidate numbers. These should include reference to:

- any general difficulties encountered in preparation of materials;
- difficulties due to faulty apparatus or materials;
- accidents to apparatus or materials;
- assistance with respect to colour blindness.

Other cases of hardship, e.g. illness, temporary disability, should be reported directly to Cambridge on the Special Consideration Form.

Report on any difficulties experienced by candidates.

Declaration (to be signed by the Supervisor)

The preparation of this practical examination has been carried out so as to maintain fully the security of the examination.

Signed

Name (in block capitals) (Supervisor)

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