

Cambridge International Examinations

Cambridge International Advanced Subsidiary and Advanced Level

CHEMISTRY 9701/36

Paper 3 Advanced Practical Skills 2

October/November 2016

CONFIDENTIAL INSTRUCTIONS



Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly.

The Supervisor's attention is drawn to the form on page 7 which must be completed and returned with the scripts.

If you have any problems or queries regarding these Instructions, please contact CIE

by email: info@cie.org.uk, by phone: +44 1223 553554, by fax: +44 1223 553558,

stating the Centre number, the nature of the query and the syllabus number quoted above.

This document consists of 8 printed pages.



Safety

Supervisors are advised to remind candidates that **all** substances in the examination should be treated with caution.

Only those tests described in the question paper should be attempted. Please also see under 'Apparatus' on the use of pipette fillers, safety goggles and plastic gloves.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out.

Attention is drawn in particular, to certain materials used in the examination. The following codes are used where relevant.

C corrosive
 HH health hazard
 F flammable
 MH moderate hazard acutely toxic
 O oxidising

N hazardous to the aquatic environment

The attention of Supervisors is drawn to any local regulations relating to safety and first-aid.

'Hazard Data Sheets', relating to materials used in this examination, should be available from your chemical supplier.

Before the Examination

1 Access to the question paper is NOT permitted in advance of the examination.

2 Preparation of materials

Where quantities are specified for each candidate, they are sufficient for the experiments described in the question paper to be completed.

In preparing materials, the bulk quantity for each substance should be increased by 25% as spare material should be available to cover accidental loss. More material may be supplied if requested by candidates, without penalty.

All solutions should be bulked and mixed thoroughly before use to ensure uniformity.

Every effort should be made to keep the concentrations accurate to within one part in two hundred of those specified.

Supervisors are asked to carry out any confirmatory tests given on page 4 to ensure the materials supplied are appropriate.

If the concentrations differ slightly from those specified, the Examiners will make the necessary allowance. They should be informed of the exact concentrations.

3 Labelling of materials

Materials must be labelled as specified in these Instructions. Materials with an **FB** code number should be so labelled **without** the identities being included on the label. Where appropriate the identity of an **FB** coded chemical is given in the question paper itself.

4 Identity of materials

It should be noted that descriptions of materials given in the question paper may not correspond with the specifications in these Instructions. The candidates must assume the descriptions given in the question paper.

5 Size of group

In view of the difficulty of the preparation of large quantities of solution of uniform concentration, it is recommended that the maximum number of candidates per group be 30 and that separate supplies of solutions be prepared for each group.

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Apparatus

- 1 In addition to the fittings ordinarily contained in a chemical laboratory, the apparatus and materials specified below will be necessary.
- **2** Pipette fillers (or equivalent safety devices), safety goggles and disposable gloves should be used where necessary.
- 3 For each candidate
 - $1 \times 250 \, \text{cm}^3$ beaker
 - 1 × foamed plastic (polystyrene) cup
 - 1 × 25 cm³ measuring cylinder
 - $1 \times \text{thermometer } (-10 \,^{\circ}\text{C to } +110 \,^{\circ}\text{C at } 1 \,^{\circ}\text{C})$
 - 1 × stop clock or sight of a clock with a seconds display
 - $1 \times glass rod$
 - $1 \times \text{spatula}$
 - $1 \times 15 \, cm^3$ crucible without lid
 - $1 \times \text{tripod}$
 - $1 \times gauze$
 - $1 \times \text{pipe-clay triangle}$
 - 1 × crucible tongs
 - 1 × Bunsen burner
 - 1 × heatproof mat
 - 1 × test-tube holder
 - $8 \times test-tube*$
 - 2 × boiling tube*
 - 1 × test-tube rack
 - 2 × teat/dropping pipette
 - $1\times$ wash bottle containing distilled water pen suitable for marking glassware paper towels

access to a balance weighing to at least 0.1 g

Where balance provision is limited, some candidates should be instructed to start the examination with Question 3. See page 58 of the current syllabus for balance: candidate ratio.

^{*}Candidates are expected to rinse and reuse test-tubes and boiling tubes where possible. Additional test-tubes should be available.

Chemicals Required

- It is especially important that great care is taken that the confidential information given below does not reach the candidates either directly or indirectly.
- It should be noted that descriptions of materials given in the question paper may not correspond with the specifications in these instructions. 2
- 3 Particular requirements

hazard per candidate identity (hazards given in this column a carbon at label) identity [MH][N] FB 1 35 cm³ 1.0 mol dm⁻³ copper(II) sulfate Dissolve 249.5g of CuSO₄.5H₂O [MH] [FI][N] FB 2 0.8 ± 0.1g zinc powder 0.8 ± 0.1g zinc powder 0.8 ± 0.1g zinc powder [F] [N] supplies [MH][N] FB 3 3.0 ± 0.2g lydrated copper(II) sulfate 3.0 ± 0.2g CuSO₄.5H₂O [MH] [N] supplies [MH] FB 4 10 cm³ 0.50 moldm⁻³ methanoic acid Dilute 21 cm³ conc. HCO₂H (90%) [C] [MH] FB 5 10 cm³ 0.50 moldm⁻³ hydrochloric acid Dilute 250 cm³ of 2.0 moldm⁻³ HCl to 1 [MI] FB 6 10 cm³ mixture of ammonium chloride Papproximately 0.25 of NH₄Cl [MH] and zinc sulfate [MI] sodium carbonate 1g Approximately 0.25 of NH₄Cl [MH] and zinc sulfate Provide approximately 1 g of Na₂CO₃ IMH] container [F] magnesium strips magnesium strips provide 2 × 2 cm strips of freshly clear					
FB 1 35 cm³ 1.0 mol dm⁻³ copper(II) sulfate FB 2 0.8 ± 0.1g zinc powder FB 3 3.0 ± 0.2g hydrated copper(II) sulfate FB 4 10 cm³ 0.50 mol dm⁻³ methanoic acid FB 5 10 cm³ 0.50 mol dm⁻³ hydrochloric acid RB 5 10 cm³ 0.50 mol dm⁻³ hydrochloric acid mixture of ammonium chloride and zinc sulfate and zinc sulfate sodium carbonate 1g sodium carbonate magnesium ribbon 2 × 2 cm strips magnesium ribbon	hazard	label	per candidate	identity	notes (hazards given in this column are for the raw materials)
FB 2 0.8 ± 0.1g zinc powder FB 3 3.0 ± 0.2g hydrated copper(II) sulfate FB 4 10 cm³ 0.50 mol dm⁻³ methanoic acid FB 5 10 cm³ 0.50 mol dm⁻³ hydrochloric acid FB 6 1 g mixture of ammonium chloride and zinc sulfate sodium carbonate 1 g sodium carbonate magnesium ribbon 2 × 2 cm strips magnesium ribbon	[MH][N]	FB 1	35 cm³	1.0 mol dm-3 copper(II) sulfate	Dissolve 249.5g of $CuSO_4.5H_2O$ [MH] [N] in each dm^3 of solution.
FB 3 3.0 ± 0.2g hydrated copper(II) sulfate FB 4 10 cm³ 0.50 moldm⁻³ methanoic acid FB 5 10 cm³ 0.50 moldm⁻³ hydrochloric acid RB 6 1g mixture of ammonium chloride and zinc sulfate sodium carbonate 1g sodium carbonate magnesium ribbon 2 × 2 cm strips magnesium ribbon	[F][N]	FB 2	0.8 ± 0.1g		0.8 ± 0.1 g zinc powder [F] [N] supplied in a stoppered container.
FB 410 cm³0.50 moldm⁻³ methanoic acidFB 510 cm³0.50 moldm⁻³ hydrochloric acidFB 61gmixture of ammonium chloride and zinc sulfatesodium carbonate1gsodium carbonatemagnesium ribbon2 × 2 cm stripsmagnesium ribbon	[MH][N]	FB 3	$3.0\pm0.2\mathrm{g}$	hydrated copper(II) sulfate	3.0 ± 0.2 g CuSO $_4.5H_2$ O [MH] [N] supplied in a stoppered container.
FB 5 10 cm³ 0.50 moldm⁻³ hydrochloric acid mixture of ammonium chloride and zinc sulfate and zinc sulfate 1 g sodium carbonate 1 sodium carbonate strips strips strips strips	[MH]	FB 4	10 cm³	0.50 moldm ⁻³ methanoic acid	Dilute 21 cm ³ conc. HCO ₂ H (90%) [C] [F] in each dm ³ of solution.
FB 6 sodium carbonate magnesium ribbon 19 mixture of ammonium chloride and zinc sulfate and zinc sulfate sodium carbonate sodium carbonate strips strips strips		FB 5	10 cm³	0.50 moldm ⁻³ hydrochloric acid	Dilute 250 cm ³ of 2.0 mol dm ⁻³ HCl to 1 dm ³ with distilled water. See preparation instructions on page 56 of the current syllabus.
sodium carbonate1gsodium carbonatemagnesium2 × 2 cmmagnesium ribbonribbonstrips	[C][MH] [N]		19	mixture of ammonium chloride and zinc sulfate	Approximately 0.25g of NH ₄ C l [MH] and 0.75g of ZnSO ₄ .7H ₂ O [C] [MH] [N] should be ground together and supplied in a stoppered container. This should be freshly prepared.
magnesium2 × 2 cmmagnesium ribbonribbonstrips	[MH]	sodium carbonate	1 g	sodium carbonate	Provide approximately 1g of Na ₂ CO ₃ [MH] in a stoppered container.
	[F]	magnesium ribbon	$2\times 2\text{cm}$ strips	magnesium ribbon	Provide $2 \times 2 \text{cm}$ strips of freshly cleaned magnesium ribbon [F] .

reagents. If necessary, they may be made available from a communal supply: however, the attention of the Invigilators should be drawn to the The reagents below should also be provided. Unless otherwise stated, each candidate should require no more than 10 cm³ of any of these fact that such an arrangement may lead to contamination of reagents and enhance the opportunity for malpractice between candidates.

hazard	label	notes
	dilute hydrochloric acid	
[C]	dilute nitric acid	
[MH]	dilute sulfuric acid	
	aqueous ammonia	
[2]	aqueous sodium hydroxide	See identity details and preparation instructions on pages 56 and 57 of the current syllabus.
	0.1 mol dm ⁻³ barium chloride or 0.1 mol dm ⁻³ barium nitrate	
Z	0.05 moldm-3 silver nitrate	
[MH]	limewater	
[MH]	aqueous acidified potassium manganate(VII)	

5 The following materials and apparatus should be available.

red and blue litmus papers, aluminium foil for testing nitrate/nitrite, wooden splints and the apparatus normally used in the Centre for use with limewater in testing for carbon dioxide

Responsibilities of the Supervisor during the Examination

1 The Supervisor, or other competent chemist, must, out of sight of the candidates, carry out the experiments in Question 1 and Question 2 and complete tables of readings on a spare copy of the question paper which should be labelled 'Supervisor's Results'.

This should be done for:

each session held and each laboratory used in that session, and each batch of solutions supplied.

N.B. The question paper cover requests the candidate to fill in details of the examination session and the laboratory used for the examination.

It is essential that each packet of scripts contains a copy of the applicable Supervisor's Results as the candidates' work cannot be assessed accurately without such information.

2 The Supervisor must complete the Report Form on page 7 to show which candidates attended each session. If all candidates took the examination in one session, please indicate this on the Report Form. A copy of the Report Form must accompany each copy of the Supervisor's Results in order for the candidates' work to be assessed accurately.

The Supervisor must give details on page 8 of any particular difficulties experienced by a candidate, especially if the Examiner would be unable to discover this from the written answers.

After the Examination

Each envelope returned to Cambridge must contain the following items.

- 1 The scripts of those candidates specified on the bar code label provided.
- 2 A copy of the Supervisor's Results relevant to the candidates in 1.
- **3** A copy of the Report Form, including details of any difficulties experienced by candidates (see pages 7 and 8).
- **4** The Attendance Register.
- 5 A Seating Plan for each session/laboratory.

Failure to provide appropriate documentation in each envelope may cause candidates to be penalised.

COLOUR BLINDNESS

With regard to colour blindness it is permissible to advise candidates who request assistance on colours of, for example, precipitates and solutions (especially titration end-points). Please include with the scripts a note of the candidate numbers of such candidates.

Experience suggests that candidates who are red/green colour-blind – the most common form – do not generally have significant difficulty. Reporting such cases with the scripts removes the need for a 'Special Consideration' application.

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REPORT FORM

Thi	This form must be completed and sent to the Examiner in the envelope with the scripts.				
Се	entre number N	lame of Centre			
1	Supervisor's Results Please submit details of the readings obtained in Question 1 and Question 2 on a spare copy the question paper clearly marked 'Supervisor's Results' and showing the Centre number a appropriate session/laboratory number.				
2	The candidate numbers of candidates attending each session were:				
	First Session	Second Session			
3	The Supervisor is required to give details overleaf of any difficulties experienced by partical candidates, giving names and candidate numbers. These should include reference to:				
(a) any general difficulties encountered in preparation of materials;		aration of materials;			
	(b) difficulties due to faulty apparatus or materials;				
(c) accidents to apparatus or materials;					
	(d) assistance with respect to colour blindness				
Other cases of hardship, e.g. illness, temporary disability, should be reported direct to C		disability, should be reported direct to CIE on the			

4 A plan of work benches, giving details by candidate numbers of the places occupied by the candidates for each experiment for each session, must be enclosed with the scripts.

normal 'Application for Special Consideration' form.



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