CAMBRIDGE INTERNATIONAL EXAMINATIONS

Cambridge International General Certificate of Secondary Education

MARK SCHEME for the October/November 2015 series

0652 PHYSICAL SCIENCE

0652/22

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

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| | Gambridge 1000L Cottobernitorians 2010 | 7002 22 |
|---|---|--------------------------|
| 1 | (a) atom of an element with same number of proton/proton number; different number of neutrons/nucleon or neutron number; | [1] [1] |
| | (if no reference to 'number' 1 max, for both proton and nucleon) | |
| | (b) atomic/proton number; mass/nucleon number; | [1] [1] |
| | (c) 2 electrons in first level and 4 in outer level; | [1] |
| | | [Total 5] |
| 2 | (a) Arrow from C; vertically downwards; | [1] [1] |
| | (b) (i) <u>Use of mass \times g (= 80 \times 10); 800 N (accept use of (9.8 or 9.81 N kg⁻¹);</u> | [1] [1] |
| | (ii) Use of weight \times distance (= 800 \times 6.0) (ecf); = 4800 (N m); | [1] [1] |
| | <pre>(iii) decreases ; (moment = force × distance from X) distance decreases owtte ;</pre> | [1] [1] |
| | | [Total 8] |
| 3 | brass; graphite/sulfur; air; graphite; chlorine; | [1] [1] [1] [1] |
| | | [Total 5] |
| 4 | (a) conduction; | [1] |
| | (b) copper fastest, iron slowest; brass quicker conductor than aluminium; | [1] [1] |
| | | [Total 3] |
| | | |

Mark Scheme
Cambridge IGCSE – October/November 2015

Syllabus 0652 Paper 22

| Pa | age : | 3 | Mark Scheme | Syllabus | Paper |
|----|-------|---|--|---------------|------------|
| | | | Cambridge IGCSE – October/November 2015 | 0652 | 22 |
| 5 | (a) | (i) | loses one/an electron ; | | [1] |
| | | (ii) | C <i>l</i> ⁻; | | [1] |
| | | | 18 ; | | [1] |
| | | | | | |
| | (b) | | electrons round chlorine ; | | [1] |
| | | sh | ared pair between hydrogen and chlorine ; | | [1] |
| | | | | | |
| | (c) | | dium hydroxide/sodium oxide; | | [1] |
| | | VV | ater; | | [1] |
| | | 0 | र | | |
| | | SC | dium carbonate/hydrogencarbonate ; | | [1] |
| | | | ater AND carbon dioxide ; | | [1] |
| | | (a | ccept correct formulae) | | |
| | | (α | soopt dollade formaliae) | | |
| | | | | | [Total 7] |
| | | | | | |
| 6 | (a) | (i) | I clearly marked equal distance behind the mirror as object is in fro with the object; | nt and in lin | |
| | | | with the object, | | [1] |
| | | | (accept very small angle between incident and reflected ray, < 5°) | | |
| | | (ii) | Ray 1 correctly reflected back along its own path; | | [1] |
| | | (iii) | Ray 2 correctly reflected ; | | [1] |
| | | (iv) | normal drawn and 'r' correctly identified; | | [1] |
| | | (v) | Ray 1 correctly continued along its own path; | | [1] |
| | | | Ray 2 correctly continued along its own path; | | [1] |
| | | (vi) | E at a suitable point with between the rays | | [1] |
| | | | | | |
| | (b) | ar | gle of reflection = angle of incidence ; | | [1] |
| | ` , | | | | |
| | (c) | y virtual (accept cannot be projected onto a screen); | | | |
| | (-) | | (, | | [1] |
| | | | | | [Total 9] |
| _ | 1-1 | • | | | F41 |
| 7 | (a) | | rogen – 2 ; drogen – 8 ; | | [1] [1] |
| | | - | ygen – 4 ; | | [1] |
| | | (0 | an be listed in any order with the correct number) | | |
| | | • | ward one mark if all three names correct and no other marks gained) | | |

| | | Cambridge IGCSE – October/November 2 | 2015 | 0652 | 22 |
|---|-----|---|------------------|------|------------|
| | (b) | 0 + 12 + 48 ; 00 ; | | | [1] [1] |
| | | (correct final answer with no working scores 2 marks, ignore any unit given) | | | |
| | (c) | any number between 4 and 6.9 ; | | | [1] |
| | | (4 is acceptable but 7 is not) | | | |
| | | 7; | | | [1] |
| | | | | | [Total 7] |
| 8 | (a) | alloons are charged (by rubbing) | | | [1] |
| | | (accept charge transferred from jumper to balloon or vice versa) | | | |
| | | oth have same charge (accept both positive/negative) se charges repel; | • 1 | | [1] [1] |
| | (b) | ater conducts (charge) ; ater removes charge/balloons discharged ; | | | [1] [1] |
| | | | | | [Total 5] |
| 9 | (a) | ny one from: bloured ions/compounds/; ore than one ion formed/different oxidation states/ <i>vai</i> seful catalysts/form complexes; gh densities/melting points; | riable valencies | ·; | [1] |
| | | (accept conducts electricity or energy) | | | |
| | (b) | rsenic/selenium/bromine/krypton; | | | [1] |
| | (c) |) malachite/copper pyrites ; | | | [1] |
| | | gold/silver/mercury/platinum; | | | [1] |
| | |) unreactive ; | | | [1] |
| | | | | | |

Mark Scheme

Syllabus

Paper

| Pá | age : | 5 | Mark Scheme | Syllabus | Paper |
|----|-------|-------|--|-------------|-------------------|
| | | | Cambridge IGCSE – October/November 2015 | 0652 | 22 |
| | (d) | (i) | no reaction/no change/OWTTE; copper formed/iron dissolves/solution turns colourless; | | [1] [1] |
| | | (ii) | iron is more reactive than copper; | | [1] |
| | | | | | [Total 8] |
| 10 | (a) | (i) | heat/energy given out; | | [1] |
| | | (ii) | $2C_2H_2 + 5O_2 \rightarrow 4CO_2 + 2H_2O$; (all formulae correct – 1 mark; correct balancing – 1 mark) | | [2] |
| | (b) | (i) | carbon monoxide; | | [1] |
| | | (ii) | poisonous/toxic/prevents transport of oxygen in blood/bonds with haemoglobin; | | [1] |
| | (c) | (i) | members differ by CH ₂ /same general formula/functional group; | | [1] |
| | | | (accept similar chemical properties/physical properties increase do | own series) | |
| | | (ii) | ethane has carbon-carbon single bond; ethaene has carbon-carbon double bond; | | [1] [1] |
| | | | | | [Total 8] |
| 11 | (a) | | rect circuit diagram for fuse; cept) | | [1] |
| | (b) | = 3 | $e 	ext{ of } R = V/I \text{ (=12/3.2) };$.75 ; m or Ω ; | | [1] [1] [1] |
| | (c) | | st be greater than 3.2A (accept for 13A fuse); d smallest above 3.2A/relevant comment re 13A fuse; | | [1] [1] [1] |
| | | (If 3 | BA is chosen and reason given is that it is the nearest to current allow | w 1c). | |
| | (d) | (i) | lamp correctly drawn in parallel with the original lamp; | | [1] |
| | | (ii) | circuit current/current through fuse now larger; greater than 5A/= 6.4A; | | [1] [1] |
| | | | | | [Total 10] |

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|--------|-----|------|--|----------|------------|
| | | | Cambridge IGCSE – October/November 2015 | 0652 | 22 |
| 12 | (a) | ran | domness of decay ; | | [1] |
| | (b) | (i) | $4600 \mathrm{s}^{-1}(\mathrm{Bq})$; | | [1] |
| | | (ii) | Indication on graph of finding time at which count rate halves ; 25 $\pm2\text{s}$; | | [1] [1] |
| | (c) | Pro | tective clothing/use tongs/short exposure time/shielding etc.; | | [1] |

[Total 5]