# CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

# MARK SCHEME for the May/June 2013 series

## **5054 PHYSICS**

5054/22 Paper 2 (Theory), maximum raw mark 75

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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### Section A

1	(a)	trav	vels further in each second / in same time / between images	B1		
	(b)		(s=) d/t in any form algebraic or numerical 40 cm/s; 0.4(0) m/s			
	(c)		air resistance increases weight constant			
	(d)	or	ces balance /cancel no resultant/net force resultant of any two forces equal and opposite to third	B1	[6]	
2	(a)		ce × distance pendicular distance	M1 A1		
	(b)	(i)	$T \times 8$ or $2000 \times 2$ seen 500 N	C1 A1		
		(ii)	(two forces) equal (in magnitude) (two forces) opposite (in direction)	B1 B1	[6]	
3	(a)	(i)	( <i>W</i> =) <i>Fd</i> <b>or</b> $90 \times 0.3$ <b>or</b> $90 \times 30$	C1		
			27 J	A1		
		(ii)	(P=) W/t or Fd/t or 27(× 20)/60 or 27/3	C1		
			9(.0) W	A1		
	(b)	(i)	800 × 30/180 or 800/6 or 6 seen or proportionality clearly used	C1		
			133 <b>or</b> 130 cm	A1		
		(ii)	extension more than 143 cm <b>or</b> (extra) extension > 10 cm <b>or</b> (some) extension permanent	B1	[7]	

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4	(a)	gan	nma r	rays, visible light, infra-red		B1	
	(b)	(i)	micr	rowaves		B1	
		(ii)	sate	llite (receives and) sends/transmits/emits/boosts/am	nplifies signal	B1	
	(	(iii)	or u	er a large area over the horizon / only one (transmitten naffected by tall buildings/hills o obstructions	er/station) needed et	c. B1	[4]
5	(a)	elec	ctrons	s move onto polythene / rod		B1	
		eled	ctrons	s/negative charge move off cloth		B1	
	(b)	(reg	gion o	of space) where force is exerted on a charge		В1	
	(c)	(i)		ce charges attract rod) attracts +ve charge/ions/particles		B1	
				els like charge rod) repels –ve charge/ions/electrons/particles		B1	
		(ii)	(net)	) positive charge on water near rod		B1	[6]
6	(a)	(i)		2.1 (V) to any value between 11 and 12 (V) bove 2/2.1(V)		B1	
		(ii)	temp	perature increases / gets hotter		B1	
	(b)	(i)	(rate	e of) flow of charge/electrons		B1	
		(ii)	0.35	A cao		B1	
	(	(iii)	<b>or</b> 6	V/R algebraic /20 /0.35		C1	
			or 1	0) (A) $/R_T = 1/20 + 1/17.1$ $R_T = 9.2 (\Omega)$ seen		C1	
			0.65	5 A		A1	[7]

	Page 4	Mark Scheme	Syllabus	Paper	
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7	(a) oval/circle through or near A centered on or near X clockwise arrow on line(s) around X and none wrong				
	(b) fields (d	ue to X and Y) cancel <b>or</b> X and Y fields equal and o	pposite	B1	
	` ' ` '	he left owards X/A/B		B1	
		rent (in wire Y) and (magnetic) <u>field</u> (caused by othe wo (magnetic) fields interact	r wire)	B1	[5]
8	EITHER				
	(a) steel / m	nagnadur / alnico / magnetite		B1	
	(b) (i) me	ntion of cutting (lines of) magnetic field / change in (	magnetic) flux	M1	
	<b>or</b> f	at(est) rate of change fast(est) cutting other explanation involving time		A1	
	<b>or</b> t	tical/upright urned through 90° normal to (magnetic) field		B1	
	OR				
	(a) NOT (ga or inver	ate) ter		B1	
	<b>(b)</b> 1,0			B1	
	(c) (i) (vol	tage across R <sub>1</sub> ) <u>becomes</u> 0/low		B1	
	(ii) ded	rease any of $R_1$ , $R_2$ , $C_1$ , $C_2$		B1	[4]
				[Total:	45]

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### Section B

9	(a)	(air	) molecules hit walls / liquid (surface)	B1	
		(air	) molecules move fast(er) /great(er) kinetic energy	B1	
		•	) molecules hit <u>more</u> often/ <u>more</u> frequently/ <u>greater</u> rate / hard <u>er</u> / <u>more</u> force (liquid) molecules evaporate	B1	[3]
	(b)	(i)	(flask) <u>in</u> (pure) <u>melting</u> ice (and water)	B1	
			(flask) in (pure) boiling water / above boiling water (at one atmosphere)	B1	
		(ii)	thin(ner) tube  or large(r) flask  or more air/less liquid  or use liquid that expands more (1 mark for each)	B2	
		(iii)	divisions not equally spaced <b>or</b> scale not uniform/not proportional	C1	
		. ,	different distance (along scale) for same temperature rise  or different change in temperature for same distance (along scale)	A1	[6]
	(c)	(i)	( <i>M</i> =) $d \times V$ in any form <b>or</b> $1200 \times 5 \times 10^{-5} \times 0.15$	C1	
			$9(.0) \times 10^{-3} \text{ kg}; 0.009(0) \text{ kg}$	A1	
		(ii)	0.09(0) N ecf (i)	B1	
		(iii)	(P=) hdg in any form or (P=) F/A in any form	C1	
			1800 Pa	A1	[5]
	(d)		ids expand less (than air) great(er) forces between liquid molecules	B1	[1]
				[Total:	15]
10	(a)	cor	rect normal by eye rect angle of incidence between candidate's normal and incident ray rect angle of refraction marked between candidate's normal and BC	B1 B1 B1	[3]
	(b)	dec	crease / change in speed / wavelength	B1	[1]

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<b>(c)</b> $n = \sin i/2$		sin <i>i</i> /s	sin <i>r</i> seen in any form		C1	
	(sin <i>r</i> =) sin <b>or</b> 0.47(14)				C1	
	28(.1)°		C1	[3]		
(d	(d) refracts less at first face and on correct side of normal			B1		
		action erging	at second face away from normal so that red ray a	and blue ray are	B1	[2]
(e	e) (i)	_	e of incidence is 0 y along normal/perpendicular to glass		B1	
	(ii)		e of incidence/ $\theta$ is larger than critical angle internal reflection occurs		B1 B1	
	(iii)	reflec	cted ray drawn correctly and emerging without refra	action from block	B1	
	(iv)	<b>or</b> lig	ntually) light emerges (into air at Q) ht refracts (out at Q) veak) refracted ray appears		B1	
		_	emerging at Q coloured in some way orrect description of movement of reflected ray (as	heta decreases)	B1	[6]
					[Total:	15]
11 (a			pply with ammeter and heater in series in parallel with heater/ power supply		B1 B1	[2]
(b	o) (i)	` ,	$VI$ in any form $2 \times 12$		C1	
		50(.4	4) W		A1	
	(ii)	( <i>E</i> =)	$Pt$ i.e. any power × any time e.g. $50(.4) \times 8$		C1	
			13(3) seen vision by 1000 seen anywhere		C1	
		0.006	67(2) (kWh)		A1	[5]

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(c) (i) molecules escape (from surface/leave water) / become gas or vapour / break bonds

C1

fast(er) moving / high energy/ energetic molecules escape

Α1

(ii)

change M1	explanation A1
wind / draught / breeze	wind knocks molecules away
or larger surface area	more chance/possibility of escape/more space to escape or more molecules come to/near/at surface
or decrease humidity / drier air	fewer molecules return/from air
or decrease atmospheric pressure	fewer air molecules to hit during escape

(iii) evaporation occurs at surface and boiling inside liquid/bubbles evaporation occurs at any temperature (accept room temperature) and boiling occurs at boiling point/100°C/ fixed / specific temperature evaporation increased by draughts/higher temp/more area and boiling is not OR increase in pressure stops boiling but only reduces evaporation any two

B2 [6]

(d) water heats air (by conduction)

В1

or water loses heat/energy (to cup or air)or air gains heat/energy (from water)

hot / heated air / particles rise

B1 [2]

- or cold air / particles sink
- or hot air is less dense
- or cold air is more dense

[Total: 15]