MARK SCHEME for the October/November 2015 series

0653 COMBINED SCIENCE

0653/31

Paper 3 (Extended Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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P	age 2	2		Mark Scheme		Syllabus	Paper
			Cambridge I	GCSE – October/No	vember 2015	0653	31
1	(a)				[]		
			ciliated cells		bacteria are trapped		
		r	nillions of alveoli present		short diffusion distance		
		m c	ucus produced by		mucus is moved upwards		
		th	in walls of alveoli		large surface area for gas exchange		
		thre	ee or two correct: 2 ma	rks, one correct: 1 m	ark ;;		[2]
	(b)	(i)	more mucus ; cilia are paralysed/da	amaged ;			[2]
		(ii)	bacteria/pathogens r	emain in the mucus ;			[1]
	(c)	(i)	$C_6H_{12}O_6 + 6O_2 \rightarrow 6$ formulae correct ; equation is balanced	6CO ₂ + 6H ₂ O and single arrow sho	wn LHS to RHS ;		[2]
		(ii)	by red (blood) cells/h	aemoglobin ;			[1]
	(d)	(i)	(person C - must be p (person C had) higher when first measureme	oresent to award mar st carbon monoxide ent taken/owtte ;	<i>k)</i> concentration at 08.00 h	ours /	[1]
		(ii)	 (ii) person B; carbon monoxide level in blood greater at 14.00/17.00 hours (compared with 11.00 hours)/carbon monoxide level in blood increased during the day/ from 2.2 to 4.8; 			[2]	
							[Total: 11]

P	age	3	Mark Scheme	Syllabus	Paper
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2	(a)	in	creases ;		[1]
	(b)	(i	bromine ;		[1]
		(ii	2NaBr + $Cl_2 \rightarrow 2NaCl + Br_2$ formulae ; balancing consequential on formulae ;		[2]
		(iii) chlorine bromine iodine <i>(must be in this order)</i> ;		[1]
		(iv	a more reactive element/halogen displaces less reactive one/ORA fluorine most reactive ;	;	[2]
	(c)	(n el io is	egative) fluoride ions move to/attracted to (positive) anode ; ectrons move from fluoride ion onto anode ; ns are discharged/1 electron moves from fluoride ion onto anode/ lost (from each ion) ;		[max 2] [Total: 9]
3	(a)	G	weight/gravitational force/gravity:		[1]
-	()	(ii	arrow pointing vertically upwards ;		[1]
	(b)	(i	tick in first box ;		[1]
		(11)	speed		

line from *y*-axis with negative gradient (accept straight or curved); line meets x-axis;

[2]

Page	e 4	Mark Scheme	Syllabus	Paper
		Cambridge IGCSE – October/November 2015	0653	31
(0	;) (i) kinetic potential ;		[1]
	(ii) (potential energy transferred =) mgh or $80 \times 10 \times 40$; = 32000 (J);		[2]
				[Total: 8]
4 (a	a) (i) cell wall correctly labelled ; (large) vacuole correctly labelled ;		[2]
	(ii) (in either order) (cell wall) provides support (for the cell) ; (large vacuole) contains cell sap/correct named nutrient (for storag provides support/shape inside the cell ;	e)/	[2]
(k) (i) leaf X has a smaller area than leaf Y /leaf X has deeper lobes/owthe	е;	[1]
	(ii) smaller area gives less water loss ; by transpiration ; OR		
		deeper lobes allow more light through/owtte ; for photosynthesis in lower leaves ;		[max 2]
	(iii) larger area for trapping light ; for <u>photosynthesis</u> ;		[2]
				[Total: 9]
				• •

5 (a) natural gas/biogas/other correct;



all 4 correct = 2 marks, 3 or 2 correct = 1 mark ;;

(ii) <u>chemical</u> (potential) *to* thermal (heat)/light/sound/kinetic ; [1]

[1]

[2]

(iii) exothermic; [1]

Page 5		Mark Scheme	Syllabus	Paper
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(c)	(i)	(2 because) in Period 2 ; (6 because) in Group VI/6 ; <i>(allow explanations based on the electron configuration 2,6)</i>		[2]
((ii)	4 shared pairs ; correct symbols and all else correct ;		[2]
(d)	(i)	noble/inert gases/Group 0 or 8/Group VIII ;		[1]
((ii)	all/outer shells complete/filled ;		[1]
				[Total: 11]

 6 (a) (i) ray in glass bent towards normal ; emergent ray parallel to incident ; angles of incidence and refraction shown correctly ;



 (ii) ray from Sun bending towards normal on entering atmosphere and reaching X ;

Sun's rays are refracted (by the atmosphere);

Sun -



[2]

[3]

(b) (i) infra-red/IR ;[1](ii) sand is better absorber of infra-red/radiation than (sea) water ;[1](c) use of v = $f\lambda$;
 λ decreases ;[2]

Page 6		6	Mark Scheme Syllabus	
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7	(a)	or de	organisms that feed on/get their energy from/reference to respiration ; lead or waste organic matter ;	
	(b)	(i)	enzymes break down the wood/large molecules into small molecules ; that can be absorbed (by the fungi) ;	[2]
		(ii)	may slow down/stop process ; due to denaturation of digestive enzymes ;	[2]
				[Total: 6]
8	(a)	(i)	(rate of reaction decreases due to) decreasing concentration/ORA;	[1]
		(ii)	X vertically in line with 8–9 time units ;	[1]
		(iii)	acid used up ;	[1]
	(b)	(i)	increased initial value on vertical axis ; intercept with time axis before 8 minutes ;	[2]
		(ii)	particles move/collide faster/have more <u>kinetic</u> energy ; collide more frequently ; greater chance of reaction during collision/owtte ;	
			(accept answers referring to activation energy)	[max 2]
				[Total: 7]
9	(a)	(i)	(R =) $\frac{V}{I}$ or $\frac{1.2}{0.5}$;	
			$= 2.4 (\Omega);$	[2]
		(ii)	1.2 (Ω) (ecf) ;	[1]
	(b)	(i)	P = IV ;	[1]
		(ii)	watt and W ;	[1]
		(iii)	(energy =) power × time or 1.2 × 0.5 × 120 ; = 72 (J) ;	[2]
	(c)	(i)	convection ;	[1]
		(ii)	by conduction ;	
			reference to particles in wire vibrating more quickly ; reference to vibrational collisions (between resistance and connecting wires) ; (also allow answers discussing the role of delocalised electrons)	[max 2]
				[Total: 10]