

Cambridge International Examinations Cambridge International General Certificate of Secondary Education

## **COMBINED SCIENCE**

0653/33 October/November 2016

Paper 3 Extended Theory MARK SCHEME Maximum Mark: 80

Published

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Ρ	age 2	2	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – October/November 2016	0653	33
1		(i) (ii)	work done = force × distance / F × d / 300 × 0.4 ; = 120 ; J ; 120 (J) ;		[3]
		. ,	allow ecf from (i)		[1]
	(b)	(i)	KE = $\frac{1}{2} \text{ mv}^2 / \frac{1}{2} \times 0.1 \times (30)^2$ ; = 45 (J);		[2]
		(ii)	efficiency = energy out/energy in × 100 <i>(or equivalent)</i> ;		
			= (45/120) × 100 = 37.5 (%);		[2]
		(iii)	(slowed down by) air resistance/friction/other reasonable opposin	g force ;	[1]
2	(a)	(i)	stopwatch/timer;		[1]
		(ii)	calcium/Ca ;		[1]
		higi (so <i>ten</i> Hig par (so mo cale ma	<i>incentration</i> her concentration increases speed/less time for the reaction ; her number of particles/particles are closer together ; ) particles collide more often ; <i>inperature</i> her temp increases speed/less time for the reaction ; ticles have more energy/are moving faster ; ) particles collide more often / particles collide with more energy/ re successful collisions ;		[max 4]
	(d)	all (tes			[1]
		(tes	st-tube <b>B</b> ) lead is more reactive than copper ;		

(test-tube  ${\boldsymbol{\mathsf{B}}})$  lead is more reactive than copper ;

Page 3		Mark Scheme	Syllabus	Paper
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3	(a) (i)	blood passes through the heart twice for each circuit/there are two circulation paths – one to the lungs and one to the body ;		[1]
	(ii)	<b>C and</b> the aorta ; takes blood from left ventricle/chamber with the thickest wall/blood has to go greater distance / owtte ;	Ł	[2]
	(iii)	(valves) prevent backflow of blood ; (wide lumen) reduces resistance of blood flow ;		[2]
	(b) (i)	to supply more oxygen / glucose (to the cells / muscles) ;		
		for respiration ; to release / supply more energy (for contraction of muscles) ; to remove carbon dioxide more quickly ;		[max 3]
	(ii)	any suitable activity, e.g. walking <b>and</b>		
		activity is more energetic / active / uses more oxygen than sitting b energetic / active / uses less oxygen than running ;	ut is less	[1]
4	(a) (i)	Gamma ultra-violet infra-red radio waves		
		infra-red ;		
		in correct box ;		[2]
	(ii)	All e/m waves travel at same (high) speed ( <i>in vacuo</i> ) ;		[1]
	(b) (i)	(matt) black ; <i>accept</i> reasonable alternatives that have a deep hue,	o a deen	
	(6) (1)	(or dark) blue black is a better absorber of i/r ;	e.g. <u>ueep</u>	[2]
	(!!)			[]
	(ii)	conduction ; convection ;		[2]
	(iii)	water expands on heating ; heated/hot/warm water less dense than cold, (so rises while cold ; ; <i>owtte</i>	sinks)	[2]

P	age 4	1	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – October/November 2016	0653	33
5	(a)	(i)	methane;		[1]
		(ii)	fossil/non-renewable ; <u>fractional</u> distillation ;		[2]
	(b)	C <sub>8</sub> H C	H <sub>18 ;</sub> 2H <sub>4 ;</sub>		[2]
	(c)	(i)	contains carbon & hydrogen ; <i>and one of</i> a compound / molecule ;		
			only ;		[2]
		(ii)	$C_6H_{12}$ because (general formula is) $C_nH_{2n}/unsaturated/contains a double bond ;$		[1]
		(iii)	bromine / bromine water ; (alkanes) no change and (alkenes) decolourised ;		[2]
6	(a)	lon (sti	amen) ger/found dangling outside the flower ; gma) thery/larger/found outside the flower ;		[2]
	(b)	sma	e 1 – no mark aller/lighter therefore can be carried by the wind ; oother surface therefore less friction/air resistance ;		[max 1]
	(c)	(i)	any two from increased rate of transpiration (at 27 °C) / more water lost from plar molecules have more (kinetic) energy ;	nt ;	[2]
		(ii)	any value less than 1.1 cm because the rate of evaporation/transp lower in humid conditions ;	iration is	[1]
	(d)	it h	t <b>X</b> – no mark as <u>root hair</u> cells ; ger surface area for absorption of water ;		[2]
	(e)		os light energy ; averts it to chemical energy /glucose ;		[2]

Page 5			Mark Scheme	Syllabus	Paper
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7	(a)	(i)	50 (cm) ;		[1]
		(ii)	maximum displacement (from no displacement to peak);		[1]
	(b)		variable resistor symbol; ammeter symbol; all connected in series to form a complete circuit;		[3]
		.,	$R = V \div I / 3 \div 2;$ = 1.5 (\Omega); power = V × I/3 × 2; = 6 watt(s) / W;		[2]
8	(a)		d at room temp/below melting point ; s must be mobile ;		[2]
	(b)	ion	s move towards the anode/positive electrode ; s lose (two) electrons/(two) electrons move to the anode/ions are charged/become atoms ;		[2]
	(c)		uced because it loses oxygen ; ept aluminium (ions) gain electrons		[1]
	(d)	2 e	lectrons in 1st shell and 8 in 2nd shell ;		[1]
	(e)		ic ; tal and non-metal combined ;		[2]

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9 (a)



(b) (i) burning fossil fuels / deforestation ;

(ii)	(explanation of greenhouse gas) absorbs heat/infra-red radiation from the earth ; causes the temperature of the atmosphere to rise/global warming carbon dioxide is a greenhouse gas ; consequence, e.g. flooding/melting ice caps/changes in weather patterns/avp ;
	Pone

[3]

[1]

[3]