

## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS General Certificate of Education Ordinary Level

CHEMISTRY 5070/03

Paper 3 Practical Test

October/November 2009

**CONFIDENTIAL INSTRUCTIONS** 

Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly.

If you have any problems or queries regarding these Instructions, please contact CIE

by e-mail: International@cie.org.uk,

by phone: +44 1223 553554, by fax: +44 1223 553558,

stating the Centre number, the nature of the query and the syllabus number quoted above.

The Supervisor's attention is drawn to the form on page 7 which must be completed and returned with the scripts.



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## Safety

Supervisors are advised to remind candidates that **all** substances in the examination should be treated with caution. Only those tests described in the question paper should be attempted. Please also see under 'Apparatus' on the use of pipette fillers and safety goggles.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out.

Attention is drawn, in particular, to certain materials used in the examination. The following codes are used where relevant.

C = corrosive substance F = highly flammable substance

**H** = harmful or irritating substance **O** = oxidising substance

T = toxic substance N = dangerous for the environment

The attention of Supervisors is drawn to any local regulations relating to safety and first-aid.

'Hazard Data Sheets', relating to materials used in this examination, should be available from your chemical supplier.

## **Preparing the Examination**

1 Access to the question paper is NOT permitted in advance of the examination.

## 2 Preparation of materials

Where quantities are specified for each candidate, they are sufficient for the experiments described in the question paper to be completed.

In preparing materials, the bulk quantity for each substance should be increased by 25% as spare material should be available to cover accidental loss. More material may be supplied if requested by candidates, without penalty.

All solutions should be bulked and mixed thoroughly before use to ensure uniformity.

Every effort should be made to keep the concentrations accurate within one part in 50 of those specified.

Supervisors are asked to carry out any confirmatory tests given on page 4 to ensure the materials supplied are appropriate.

## 3 Labelling of materials

Materials must be labelled as specified in these instructions. Materials with a letter code (e.g. **P**, **Q**) should be so labelled, **without** the identities being included on the label – where appropriate, the identity of a letter-coded chemical is given in the question paper itself.

## 4 Identity of materials

It should also be noted that descriptions of solutions given in the question paper may not correspond exactly with the specification in these Instructions. **The candidates must assume the descriptions given in the question paper.** 

## 5 Size of group

In view of the difficulty of the preparation of large quantities of solution of uniform concentration, it is recommended that the maximum number of candidates per group be 30 and that separate supplies of solutions be prepared for each group.

## **Apparatus**

- 1 In addition to the fittings ordinarily contained in a chemical laboratory, the apparatus and materials specified below will be necessary.
- 2 Pipette fillers (or equivalents safety devices) and safety goggles should be used where necessary.
- 3 For each candidate
  - 1 x 50 cm<sup>3</sup> burette
  - 1 x 20 cm<sup>3</sup> or 25 cm<sup>3</sup> pipette

(It is essential that all candidates at a Centre have a pipette of the same capacity.)

- 1 x pipette filler
- 1 x flask or other suitable vessel for titration
- 1 x stand
- 1 x burette clamp
- 1 x funnel for filling burette
- 1 x white tile
- 1 x wash bottle of distilled water
- a supply of test-tubes
- 1 x test-tube rack
- 1 x stirring rod

## **Chemicals Required**

It is especially important that great care is taken that the confidential information given below does not reach the candidates either directly or indirectly.

# 2 Particular requirements

hazard	label	per candidate	identity		notes
Ξ	<b>a</b>	150 cm <sup>3</sup>	solution containing 5g/dm <sup>3</sup> sodium carbonate and 1g/dm <sup>3</sup> sodium chloride	nate	aqueous solution containing 5g anhydrous sodium carbonate and 1g sodium chloride per $\mbox{dm}^3$
[H] [C]	Ø	150 cm <sup>3</sup>	0.1 mol/dm <sup>3</sup> hydrochloric acid	oric acid	
			methyl orange or screened methyl orange indicator	eened tor	
Supervis	sors are a	sked to carry	out a standard acid/ba	se titratior	Supervisors are asked to carry out a standard acid/base titration between solutions P and Q to ensure that the concentrations of the two solutions
fall withir 18.0 cm <sup>3</sup>	n the give and 22.0	fall within the given range. It is 18.0 cm <sup>3</sup> and 22.0 cm <sup>3</sup> of <b>P</b> .)	fall within the given range. It is <b>essential</b> that $25.0\mathrm{cm}^3$ of $18.0\mathrm{cm}^3$ and $22.0\mathrm{cm}^3$ of <b>P</b> .)	³ of <b>Q</b> rea∈	<b>Q</b> reacts with between $23.0\mathrm{cm}^3$ and $27.0\mathrm{cm}^3$ of <b>P</b> (or $20.0\mathrm{cm}^3$ of <b>Q</b> reacts with between
Ξ	œ	10 cm <sup>3</sup>	approximately 0.2 mol/dm <sup>3</sup> zinc sulfate	ıl/dm³	58 g of hydrated zinc sulfate, ${\rm ZnSO_4.7H_2O}$ , in 1 dm $^3$ of distilled water
Ξ	ဟ	10 cm <sup>3</sup>	approximately 0.2 mol/dm <sup>3</sup> iron(II) sulfate	ol/dm <sup>3</sup>	dissolve 56 g of hydrated iron(II) sulfate, ${\rm FeSO_4.7H_2O}$ , in $500{\rm cm^3}$ of distilled water and $50{\rm cm^3}$ 1 mol/dm $^3$ sulfuric acid and dilute the solution with distilled water to 1 dm $^3$
Ε	<b>-</b>	10cm <sup>3</sup>	approximately 0.1 mol/dm <sup>3</sup> lead(II) nitrate	1/dm <sup>3</sup>	$33\mathrm{g}$ of lead(II) nitrate in 1 dm $^3$ of distilled water
Ξ	adneor	aqueous hydrogen peroxide		10 cm <sup>3</sup>	"20 volume" hydrogen peroxide solution
Ξ	adneor	us potassium	aqueous potassium manganate(VII)	10 cm <sup>3</sup>	0.02 mol/dm³ potassium manganate(VII)

The standard bench reagents specifically required are set out below. If necessary, they may be made available from a communal supply: however, the attention of the Invigilators should be drawn to the fact that such an arrangement may enhance the opportunity for malpractice between candidates. ന

hazard	label	identity	notes
[0]	dilute nitric acid	1.0 mol/dm <sup>3</sup> nitric acid	
[H]	aqueous silver nitrate	0.05 mol/dm <sup>3</sup> silver nitrate	
Ξ	aqueous barium nitrate	0.2 mol/dm³ barium nitrate	0.2 mol/dm <sup>3</sup> barium chloride (labelled barium nitrate) may be used as an alternative.
<u>ට</u>	aqueous sodium hydroxide	1.0 mol/dm <sup>3</sup> sodium hydroxide	
<u></u>	aqueous ammonia	1.0 mol/dm <sup>3</sup> ammonia	

The reagents, materials and apparatus to test the gases listed in the syllabus must be available to candidates. If necessary, they may be made available from a communal supply: however, the attention of the Invigilators should be drawn to the fact that such an arrangement may enhance the opportunity for malpractice between candidates.

hazard	label	identity	notes
Ξ	limewater	saturated aqueous calcium hydroxide, Ca(OH) <sub>2</sub>	Prepare fresh limewater by leaving distilled water to stand over solid calcium hydroxide for several days, shaking occasionally. Decant or filter the solution.
[T][N]	acidified aqueous $egin{array}{c} potassium & dichromate(\mathrm{VI}) \end{array}$	$0.05  \text{mol/dm}^3  \text{K}_2 \text{Cr}_2 \text{O}_7$ $0.05  \text{mol/dm}^3  \text{H}_2 \text{SO}_4$	Dissolve 14.8g of K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub> <b>[T][N]</b> in 50 cm <sup>3</sup> of 1 mol/dm <sup>3</sup> sulfuric acid <b>[H] [C]</b> . Make the solution up to 1 dm <sup>3</sup> with distilled water.  The use of plastic gloves may be considered to prevent contact with skin.
		red and blue litmus paper or universal indicator paper	
		plain filter strips for use with aqueous potassium dichromate $\left( VI\right)$	
		wooden splints	
		the apparatus normally used in the Centre for use with limewater in testing for carbon dioxide	

## **During the Examination**

1 The Supervisor, or other competent chemist must carry out the experiments in question 1 and question 2 and record the results on a spare copy of the question paper which should be labelled 'Supervisor's Results'.

This should be done for:

each session held and each laboratory used in that session, and each set of solutions supplied.

It is essential that each packet of scripts contains a copy of the applicable Supervisor's Results as the candidates' work cannot be assessed accurately without such information.

2 The Supervisor must complete the Report Form on page 7 to show which candidates attended each session. If all candidates took the examination in one session, please indicate this on the Report Form. A copy of the Report Form must accompany each copy of the Supervisor's Results in order for the candidates' work to be assessed accurately.

The Supervisor must give details on page 8 of any particular difficulties experienced by a candidate, especially if the Examiner would be unable to discover this from the written answers.

### After the Examination

Each envelope returned to Cambridge must contain the following items.

- 1 The scripts of those candidates specified on the bar code label provided.
- 2 A copy of the Supervisor's Report relevant to the candidates in 1.
- A copy of the Report Form, including details of any difficulties experienced by candidates (see pages 7 and 8).
- 4 The Attendance Register.
- 5 A Seating Plan for each session/laboratory.

Failure to provide appropriate documentation in each envelope may cause candidates to be penalised.

## **Colour Blindness**

With regard to colour-blindness – a minor handicap, relatively common in males – it is permissible to advise candidates who request assistance on colours of, for example precipitates and solutions (especially titration end-points). Please include with the scripts a note of the candidate numbers of such candidates.

Experience suggests that candidates who are red/green colour-blind – the most common form – do not generally have significant difficulty. Reporting such cases with the scripts removes the need for a 'Special Consideration' application for this handicap.

## **REPORT FORM**

This form must be com	pleted and sent to the	e Examiner in the env	elope with the scripts.

Centre Number	Name of Centre

## 1 Supervisor's Results

Supervisors are asked to use a spare copy of the question paper to report their results for **Q.1** and **Q.2** and enclose this copy of the question paper with the candidates' answers. This copy of the question paper should be clearly labelled 'Supervisor's Results'. Failure to enclose these results and this report form may lead to candidates being unavoidably penalised.

If candidates from more than one Centre are taking the examination, it is essential that a copy of the 'Supervisor's Results' should be sent with the scripts from **each Centre**.

2 The candidate numbers of candidates attending each session were:

First Session Second Session

		8
3		Supervisor is invited to report details of any difficulties experienced by candidates, giving nes and candidate numbers. The report should include reference to:
	(a)	any general difficulties encountered in making preparation;
	(b)	difficulties due to faulty apparatus or material;
	(c)	accidents to apparatus or materials;
	(d)	assistance with respect to colour-blindness.
		Other cases of hardship, e.g. illness, temporary disability, should be reported directly to CIE on the normal 'Application for Special Consideration' form.
4		lan of work benches, giving details by candidate numbers of the places occupied by the didates for each experiment for each session, must be enclosed with the scripts.
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		SIGNED
CEN	NTRE	E NUMBER
		ndidates' Centre number is different from the number of the Centre at which the examination was e Supervisor should write <b>both Centre numbers in the space provided</b> .
Dec	lara	tion (to be signed by the Principal).
		paration of the Practical examination has been carried out so as to maintain fully the security of nination.
		SIGNED



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NAME (in block capitals).....

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