UNIVERS			
		E INTERNATIONAL EXAMINATIONS	
PHYSICS		9702/04	
Paper 4		October/November 2005	
		1 hour	
Candidates answer on the Question Pap No Additional Materials are required.			
READ THESE INSTRUCTIONS FIRST			
Write in dark blue or blac	ck pen in the spaces pro	d name on all the work you hand in. vided on the Question Paper.	
Do not use staples, pape Answer <b>all</b> questions.	er clips, highlighters, glue	e or correction fluid.	
You may lose marks if ye At the end of the examin	nation, fasten all your wo	king or if you do not use appropriate units.	
		For Examiner's Us	
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		1 2 3 4 5	
		1 2 3 4	
		1 2 3 4 5 6	

## Data

speed of light in free space,	$c = 3.00  imes 10^8 \text{ m s}^{-1}$
permeability of free space,	$\mu_0 = 4\pi  imes 10^{-7} \ { m H \ m^{-1}}$
permittivity of free space,	$\epsilon_{0} = 8.85  imes 10^{-12} \ { m F}  { m m}^{-1}$
elementary charge,	$e = 1.60 \times 10^{-19} \text{ C}$
the Planck constant,	$h = 6.63 \times 10^{-34} \mathrm{Js}$
unified atomic mass constant,	$u = 1.66 \times 10^{-27} \text{ kg}$
rest mass of electron,	$m_{ m e} = 9.11  imes 10^{-31} \ { m kg}$
rest mass of proton,	$m_{ m p} = 1.67  imes 10^{-27} \ { m kg}$
molar gas constant,	$R = 8.31 \text{ J K}^{-1} \text{ mol}^{-1}$
the Avogadro constant,	$N_{\rm A} = 6.02 \times 10^{23}  {\rm mol}^{-1}$
the Boltzmann constant,	$k = 1.38 \times 10^{-23} \mathrm{J}\mathrm{K}^{-1}$
gravitational constant,	$G = 6.67 \times 10^{-11} \text{ N m}^2 \text{ kg}^{-2}$
acceleration of free fall,	$g = 9.81 \text{ m s}^{-2}$

## Formulae

uniformly accelerated motion,	$s = ut + \frac{1}{2}at^2$ $v^2 = u^2 + 2as$
work done on/by a gas,	$W = \rho \Delta V$
gravitational potential,	$\phi = -\frac{Gm}{r}$
simple harmonic motion,	$a = -\omega^2 x$
velocity of particle in s.h.m.,	$v = v_0 \cos \omega t$ $v = \pm \omega \sqrt{(x_0^2 - x^2)}$
resistors in series,	$R = R_1 + R_2 + \dots$
resistors in parallel,	$1/R = 1/R_1 + 1/R_2 + \dots$
electric potential,	$V = \frac{Q}{4\pi\epsilon_0 r}$
capacitors in series,	$1/C = 1/C_1 + 1/C_2 + \dots$
capacitors in parallel,	$C = C_1 + C_2 + \dots$
energy of charged capacitor,	$W = \frac{1}{2}QV$
alternating current/voltage,	$x = x_0 \sin \omega t$
hydrostatic pressure,	$p = \rho g h$
pressure of an ideal gas,	$p = \frac{1}{3} \frac{Nm}{V} < c^2 >$
radioactive decay,	$x = x_0 \exp(-\lambda t)$
decay constant,	$\lambda = \frac{0.693}{t_{\frac{1}{2}}}$
critical density of matter in the Univers	se, $\rho_0 = \frac{3H_0^2}{8\pi G}$
equation of continuity,	Av = constant
Bernoulli equation (simplified),	$p_1 + \frac{1}{2}\rho v_1^2 = p_2 + \frac{1}{2}\rho v_2^2$
Stokes' law,	$F = Ar\eta v$
Reynolds' number,	$R_{\rm e} = \frac{\rho v r}{\eta}$
drag force in turbulent flow,	$F = Br^2 \rho v^2$
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[Turn over

[3]

Answer **all** the questions in the spaces provided.

- 1 The Earth may be considered to be a sphere of radius  $6.4 \times 10^6$  m with its mass of  $6.0 \times 10^{24}$  kg concentrated at its centre. A satellite of mass 650 kg is to be launched from the Equator and put into geostationary orbit.
  - (a) Show that the radius of the geostationary orbit is  $4.2 \times 10^7$  m.

(b) Determine the increase in gravitational potential energy of the satellite during its launch from the Earth's surface to the geostationary orbit.

- energy = ..... J [4]
- (c) Suggest one advantage of launching satellites from the Equator in the direction of rotation of the Earth.

.....[1]

5

amount = ..... mol [2]

(c) The pressure in the tyre is to be increased using a pump. On each stroke of the pump, 0.012 mol of air is forced into the tyre. Calculate the number of strokes of the pump required to increase the pressure to  $3.4 \times 10^5$  Pa at a temperature of 27 °C.

(a) State the first law of thermodynamics in terms of the increase in internal energy  $\Delta U$ , the heating q of the system and the work w done on the system. ......[1] (b) The volume occupied by 1.00 mol of liquid water at 100 °C is  $1.87 \times 10^{-5}$  m<sup>3</sup>. When the water is vaporised at an atmospheric pressure of  $1.03 \times 10^5$  Pa, the water vapour has a volume of  $2.96 \times 10^{-2} \text{ m}^3$ . The latent heat required to vaporise 1.00 mol of water at 100 °C and 1.03×10<sup>5</sup> Pa is  $4.05 \times 10^4$  J. Determine, for this change of state, the work w done on the system, (i) *w* = ...... J [2] (ii) the heating q of the system, *q* = ..... J [1] (iii) the increase in internal energy  $\Delta U$  of the system.  $\Delta U = \dots J [1]$ 

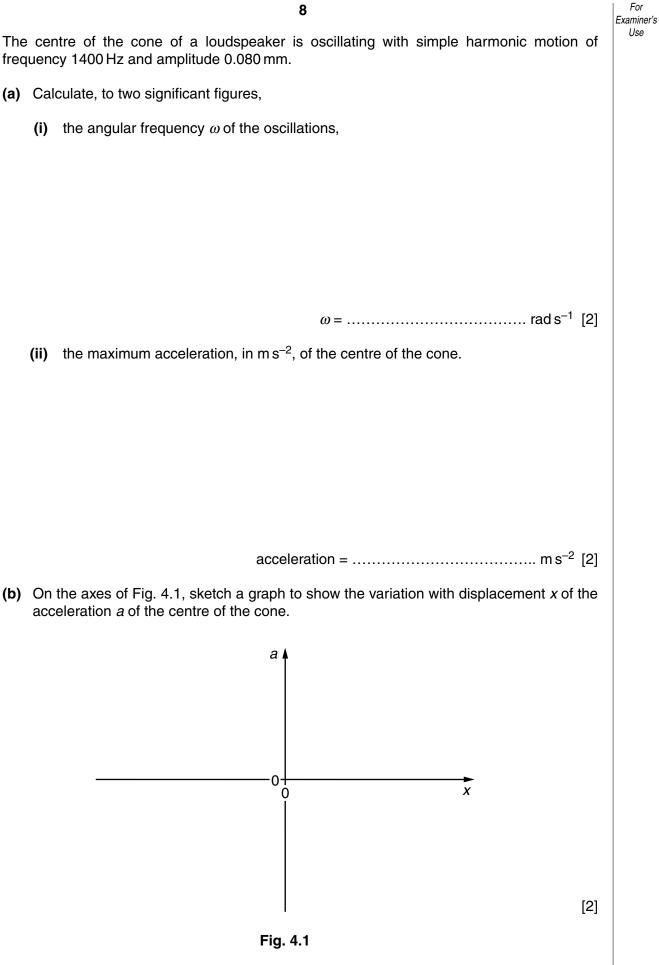
For

Examiner's Use

(c) Using your answer to (b)(iii), estimate the binding energy per molecule in liquid water.

7

energy = ..... J [2]

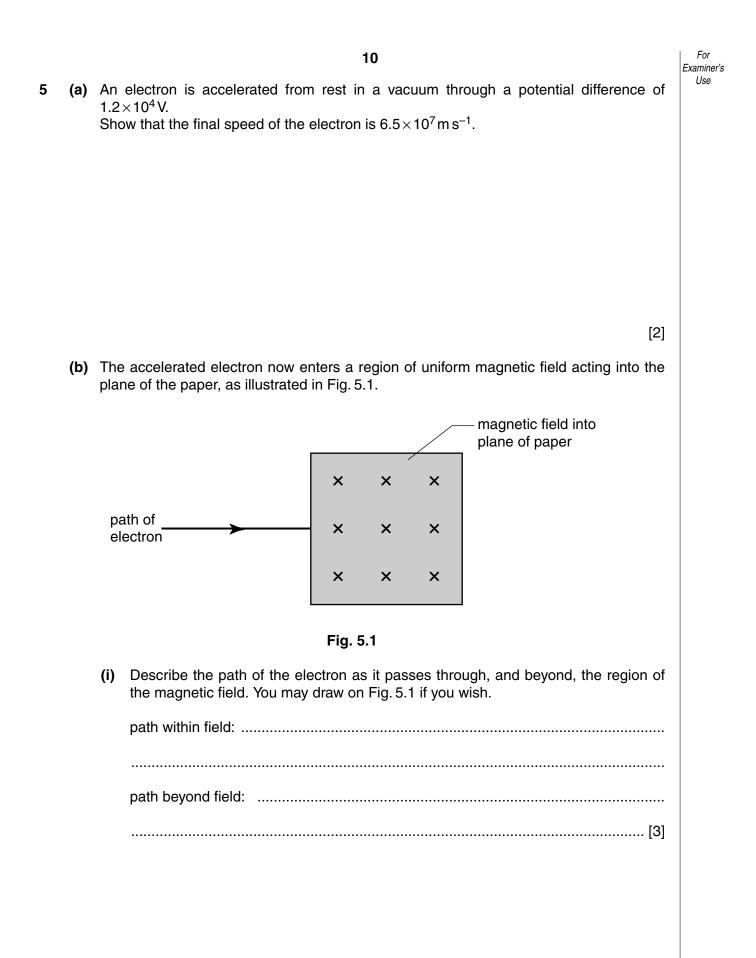


(c) (i) State the value of the displacement *x* at which the speed of the centre of the cone is a maximum.

*x* = ...... mm [1]

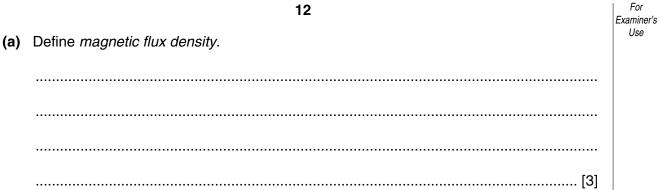
(ii) Calculate, in  $m s^{-1}$ , this maximum speed.

speed = .....  $m s^{-1}$  [2]

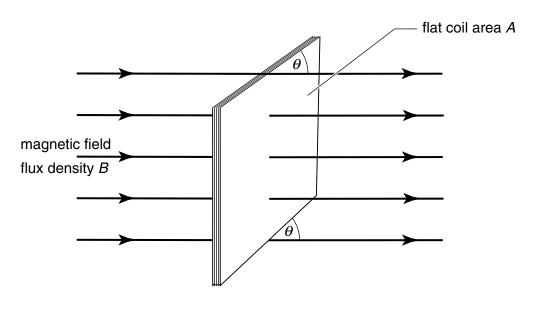


- (ii) State and explain the effect on the magnitude of the deflection of the electron in the magnetic field if, separately,
  - **1.** the potential difference accelerating the electron is reduced,

2.



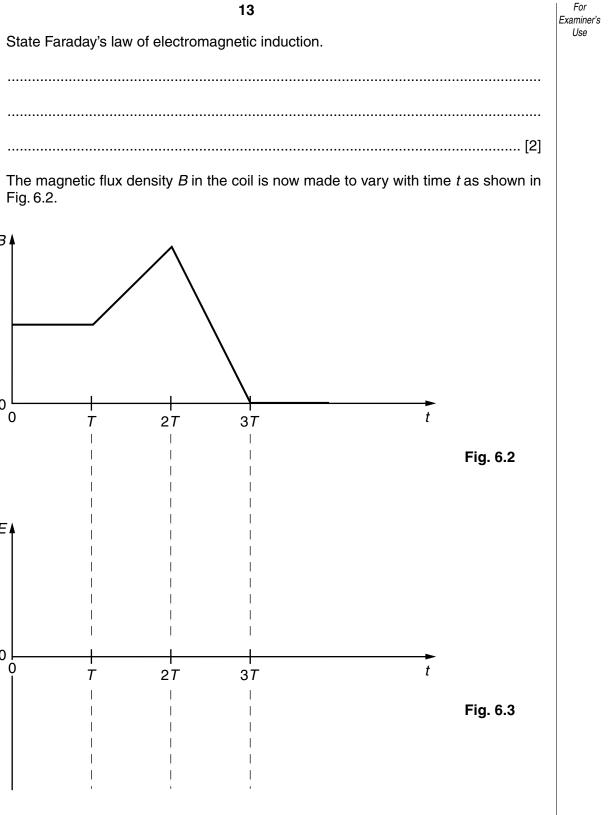
(b) A flat coil consists of *N* turns of wire and has area *A*. The coil is placed so that its plane is at an angle  $\theta$  to a uniform magnetic field of flux density *B*, as shown in Fig. 6.1.

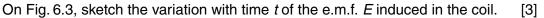




Using the symbols A, B, N and  $\theta$  and making reference to the magnetic flux in the coil, derive an expression for the magnetic flux linkage through the coil.

[2]





(c) (i)

(ii)

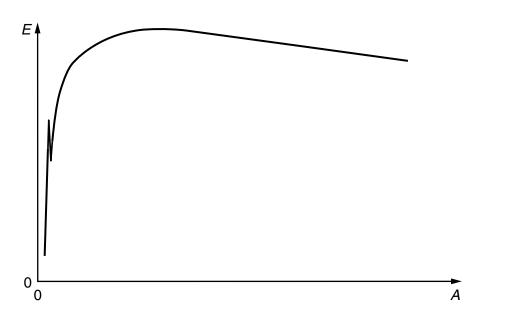
В

0

E

0 0

**7** Fig. 7.1 illustrates the variation with nucleon number *A* of the binding energy per nucleon *E* of nuclei.





(a) (i) Explain what is meant by the *binding energy* of a nucleus.

- (ii) On Fig. 7.1, mark with the letter S the region of the graph representing nuclei having the greatest stability. [1]
- (b) Uranium-235 may undergo fission when bombarded by a neutron to produce Xenon-142 and Strontium-90 as shown below.

 $^{235}_{92}\text{U} + \, {}^{1}_{0}n \rightarrow \, {}^{142}_{54}\text{Xe} + \, {}^{90}_{38}\text{Sr} + neutrons$ 

(i) Determine the number of neutrons produced in this fission reaction.

number = .....[1]

(ii) Data for binding energies per nucleon are given in Fig. 7.2.

isotope	binding energy per nucleon / MeV
Uranium-235	7.59
Xenon-142	8.37
Strontium-90	8.72



## Calculate

1. the energy, in MeV, released in this fission reaction,

energy = ..... MeV [3]

2. the mass equivalent of this energy.

mass = ..... kg [3]

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